



2008 Sulfur Dioxide Summary

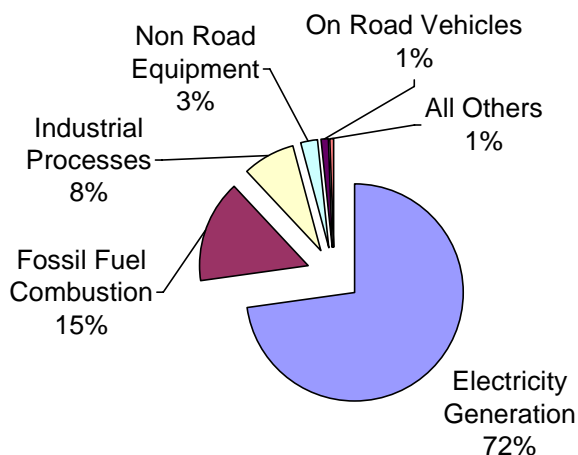
New Jersey Department of Environmental Protection

NATURE AND SOURCES

Sulfur dioxide (SO₂) is a heavy, colorless gas with a suffocating odor that easily dissolves in water to form sulfuric acid. SO₂ gases can be formed when fuels containing sulfur are burned, or when gasoline is extracted from oil. Most of the sulfur dioxide released into the air comes from electric utilities, especially those that burn coal with high sulfur content. Sulfur is found in raw materials such as crude oil, coal, and ores that contain metals such as aluminum, copper, zinc, lead and iron. Industrial facilities that derive their products from these materials may also release SO₂. A pie chart summarizing the major sources of SO₂ is shown in Figure 1.

Figure 2 (page 2) shows that SO₂ concentrations in New Jersey are generally higher in the winter than in the summer due to higher emissions from space heating and other sources. As shown in Figure 3 (page 2), SO₂ levels tend to peak in mid to late morning as emissions accumulate prior to being more effectively dispersed when wind speeds increase and atmospheric mixing increases later in the day.

Figure 1
National Summary
SO₂ Emissions by Source Sector



Source: USEPA website
[HTTP://WWW.EPA.GOV/AIR/EMISSIONS/SO2.HTM](http://www.epa.gov/air/emissions/so2.htm)
2005 data.

HEALTH AND ENVIRONMENTAL EFFECTS

Sulfur dioxide causes irritation of the mucous membranes. This is probably the result of the action of sulfurous acid that is formed when the highly soluble SO₂ dissolves at the surface of the membranes. Groups that are especially susceptible to the harmful health effects of SO₂ include children, the elderly, and people with heart or lung disorders such as asthma. When SO₂ concentrations in the air become elevated, people belonging to these sensitive groups and those who are active outdoors may have trouble breathing. The International Agency for Research on Cancer (IARC) evaluated SO₂ and based on available information, determined that no conclusion can be made as to the carcinogenicity of SO₂ to human beings.

Sulfur dioxide reacts with other gases and particles in the air to form sulfates that can be harmful to people and the environment. Sulfate particles are the major cause of reduced visibility in the eastern United States. SO₂ can also react with other substances in the air to form acids that fall to the earth in rain and snow. Better known as acid rain, this acidic precipitation can damage forests and crops, can make lakes and streams too acidic for fish, and eventually speeds up the decay of building materials and paints.

STANDARDS

There are three National Ambient Air Quality Standards (NAAQS) for SO₂. There is an annual average health standard of 0.03 parts per million (ppm). This is based on a calendar year average of continuously monitored levels. There is also a 24-hour average health based standard of 0.14 ppm which is not to be exceeded more than once a year, and a secondary (welfare based) standard of 0.5 ppm, 3-hour average concentration that is also not to be exceeded more than once per year.

New Jersey has also set state air quality standards for SO₂. They are similar to the federal standards but are

expressed in micrograms per cubic meter ($\mu\text{g}/\text{m}^3$) instead of ppm. They are also based on rolling averages rather than block averages. So, for example, the state's primary 12-month standard is based on any twelve-month average recorded during the year, while the

federal standard is based solely on the calendar year average. The state also has secondary 12-month, 24-hour, and 3-hour average standards. Table 1 summarizes the NAAQS and the New Jersey Ambient Air Quality Standards (NJAAQS) for SO_2 .

Figure 2
Sulfur Dioxide – New Jersey
2008 Monthly Variation
Parts Per Million (ppm)

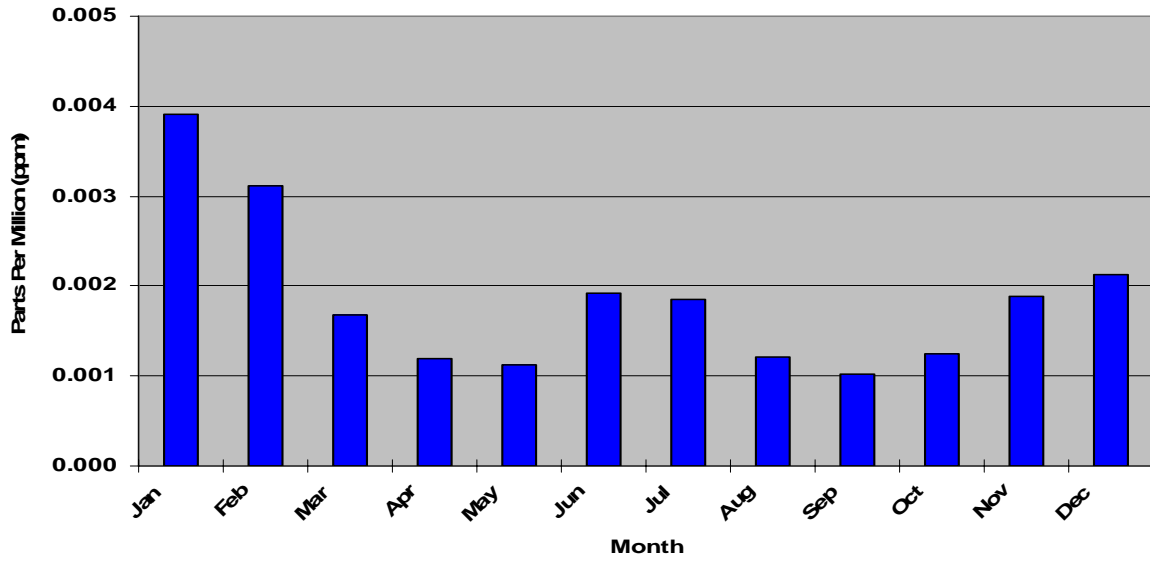


Figure 3
Sulfur Dioxide – New Jersey
2008 Hourly Variation
Parts Per Million (ppm)

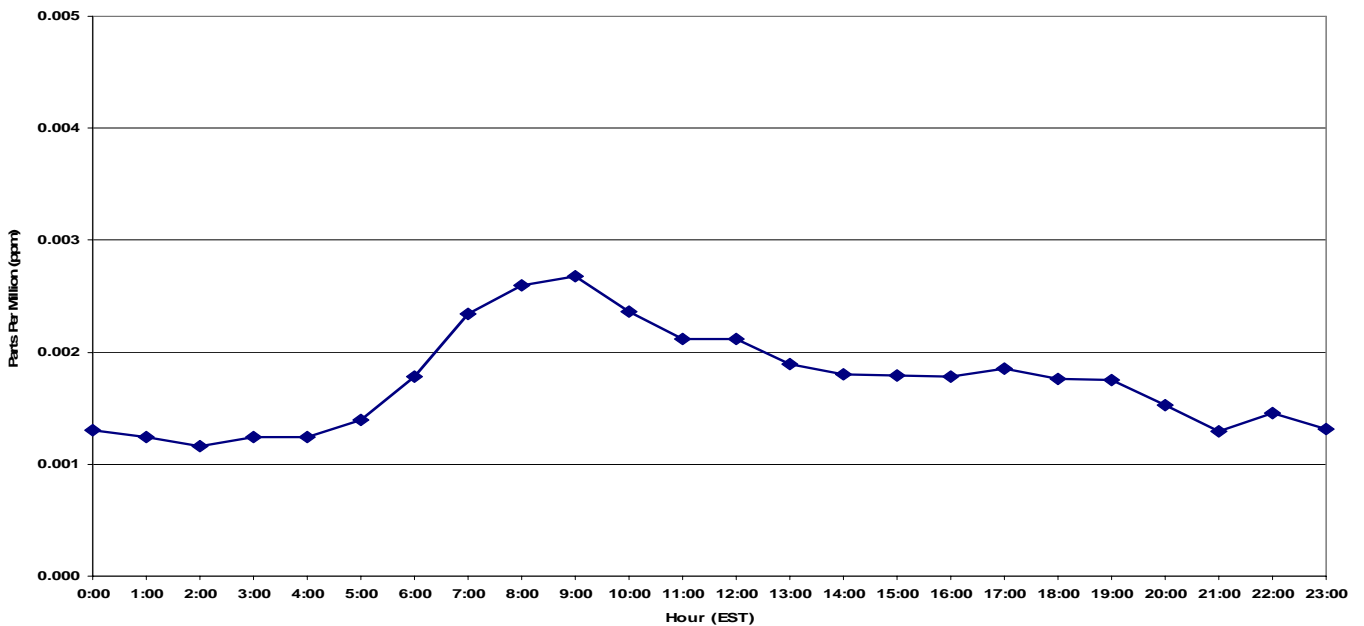


Table 1
National and New Jersey Ambient Air Quality Standards for Sulfur Dioxide
Parts Per Million (ppm)
Micrograms Per Cubic Meter ($\mu\text{g}/\text{m}^3$)

Averaging Period	Type	New Jersey	National^a
12-month average	Primary	80 $\mu\text{g}/\text{m}^3$ (0.03 ppm)	0.03 ppm
12-month average	Secondary	60 $\mu\text{g}/\text{m}^3$ (0.02 ppm)	---
24-hour average	Primary	365 $\mu\text{g}/\text{m}^3$ (0.14 ppm)	0.14 ppm
24-hour average	Secondary	260 $\mu\text{g}/\text{m}^3$ (0.10 ppm)	---
3-hour average	Secondary	1300 $\mu\text{g}/\text{m}^3$ (0.5 ppm)	0.5 ppm

^a – National standards are block averages rather than moving averages.

MONITORING LOCATIONS

The state monitored SO₂ levels at 13 locations in 2008. These sites are shown in Figure 4. Monitoring location changes included a temporary shut down of the Elizabeth site from March 7, 2007 through July 1, 2008 when major renovations were completed at the site. The Camden Lab station was shut down on September 29, 2008 because the NJDEP lost access to the site. The NJDEP is actively pursuing the establishment of a new monitoring station in Camden. The last data interruption took place at the Bayonne site, where the SO₂ data from November 2007 through March 13, 2008 was flagged as invalid because of instrument malfunction.

SO₂ LEVELS IN 2008

None of the monitoring sites recorded exceedances of the primary or secondary SO₂ standards during 2008. The maximum 12-month average concentration recorded was 0.004 ppm in Jersey City and at Elizabeth Lab. The maximum 24-hour average level recorded was 0.035 ppm. This average was recorded at Camden Lab. The highest 3-hour average recorded was 0.071 ppm at Chester. Summaries of the 2008 data are provided in Tables 2 and 3 (page 4), and Figures 5 and 6 (page 5).

Figure 4
2008 Sulfur Dioxide
Monitoring Network

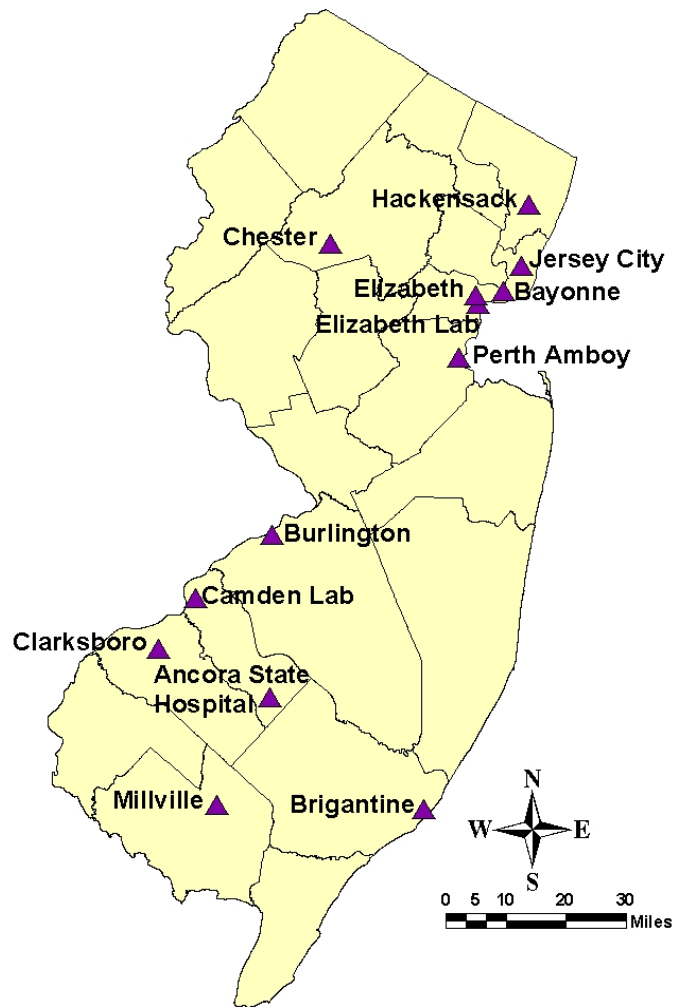


Table 2
2008 Sulfur Dioxide Data
3-Hour and Annual Averages
Parts Per Million (ppm)

Monitoring Sites	3-Hour Average Maximum	3-Hour Average 2nd Highest^b	12-Month Average Maximum	Calendar Year Average
Ancora State Hospital	0.020	0.019	0.002	0.001
Bayonne (c)	0.027	0.027	0.003	0.003
Brigantine	0.017	0.015	0.001	0.001
Burlington	0.032	0.031	0.002	0.002
Camden Lab (d)	0.041	0.038	0.003	-----
Chester	0.071	0.026	0.002	0.001
Clarksboro	0.025	0.023	0.003	0.002
Elizabeth (e)	0.018	0.014	-----	-----
Elizabeth Lab	0.040	0.030	0.004	0.003
Hackensack	0.017	0.016	0.002	0.001
Jersey City	0.030	0.027	0.004	0.003
Millville	0.021	0.017	0.002	0.001
Perth Amboy	0.034	0.018	0.002	0.002

^b – Based on non-overlapping 3 – hour moving averages.

^c – Data from November 2007 through March 13, 2008 flagged as invalid because of instrument malfunction.

^d – Station was shut down September 29, 2008 because the NJDEP lost access to the station.

^e – Temporary shut down of site from March 7, 2007 to July 1, 2008 for renovations.

Table 3
2008 Sulfur Dioxide Data
24-Hour and Daily Averages
Parts Per Million (ppm)

Monitoring Sites	24-Hour Average Maximum	24-Hour Average 2nd Highest^b	Daily Average Maximum	Daily Average 2nd Highest
Ancora State Hospital	0.009	0.008	0.008	0.008
Bayonne (c)	0.015	0.010	0.015	0.010
Brigantine	0.010	0.008	0.009	0.007
Burlington	0.013	0.011	0.012	0.010
Camden Lab (d)	0.035	0.015	0.023	0.012
Chester	0.011	0.011	0.011	0.010
Clarksboro	0.011	0.011	0.010	0.009
Elizabeth (e)	0.009	0.006	0.007	0.005
Elizabeth Lab	0.012	0.012	0.011	0.011
Hackensack	0.011	0.009	0.010	0.009
Jersey City	0.019	0.017	0.017	0.017
Millville	0.010	0.009	0.009	0.008
Perth Amboy	0.011	0.011	0.010	0.009

^b – Based on non-overlapping 24 – hour moving averages.

^c – Data from November 2007 through March 13, 2008 flagged as invalid because of instrument malfunction.

^d – Station was shut down September 29, 2008 because the NJDEP lost access to the station.

^e – Temporary shut down of site from March 7, 2007 to July 1, 2008 for renovations.

Figure 5
2008 Sulfur Dioxide Concentrations
Highest and 2nd Highest 24-Hour Averages
Parts Per Million (ppm)

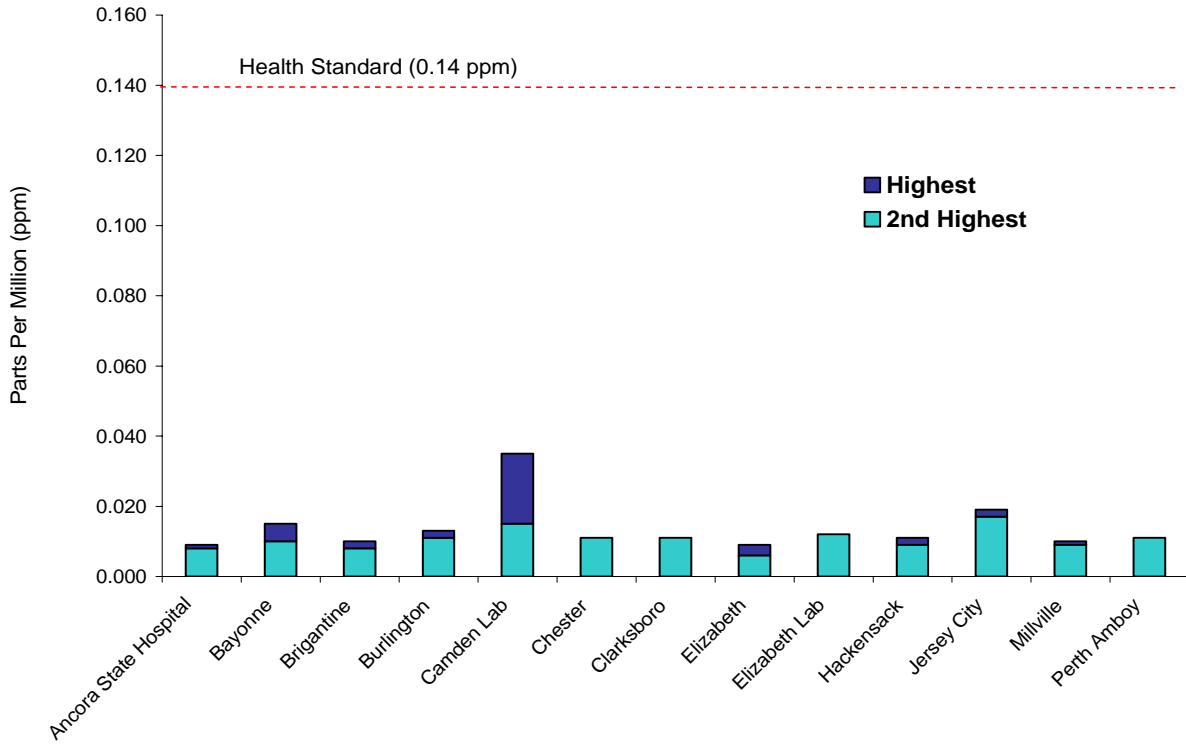
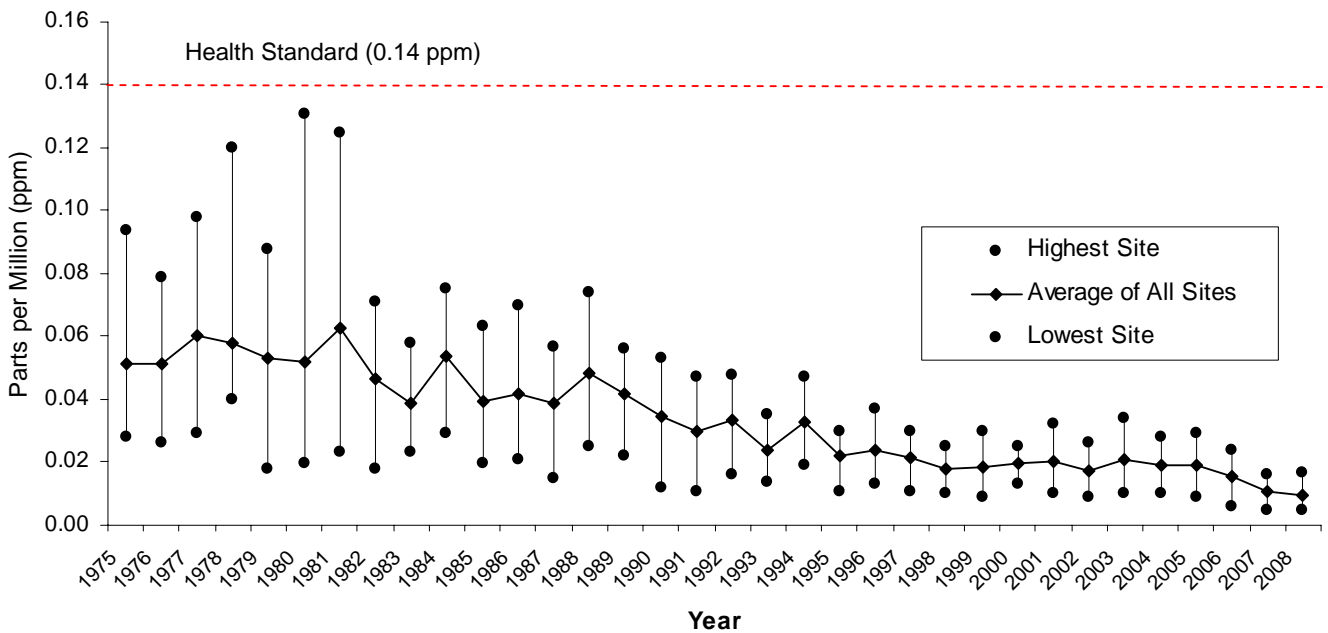


Figure 6
1975 - 2008 Sulfur Dioxide Concentrations
Second Highest Daily Average
Parts Per Million (ppm)

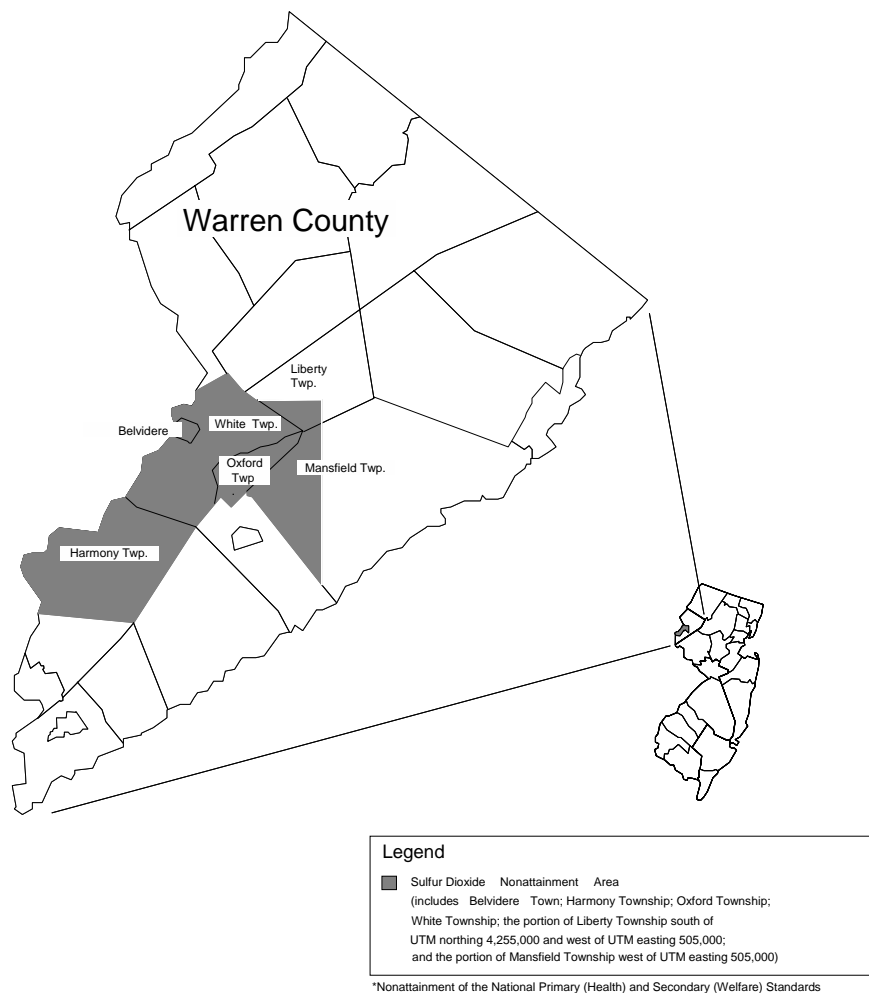


TRENDS

Since the implementation of regulations requiring the use of low sulfur fuels in New Jersey, SO₂ concentrations have improved significantly. The last time an exceedance of any of the National SO₂ standards was recorded in the state was in 1980. A trend graph of SO₂ levels showing the daily average concentrations recorded since 1975 from the highest, average, and lowest of all sites is shown in Figure 6 (page 5). The graph uses the second highest daily average, as this is the value that determines if the national health standard is being met (one exceedance per site is allowed each year).

Although there has not been a measured exceedance of the NAAQS in over two decades, there is still a small area of New Jersey that is classified as a non-attainment area for SO₂. This is the result of air quality modeling studies that predicted non-attainment of the standard within a small area of Warren County. The area is shown below in the map in Figure 7.

Figure 7
Sulfur Dioxide Non-attainment Areas in New Jersey



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